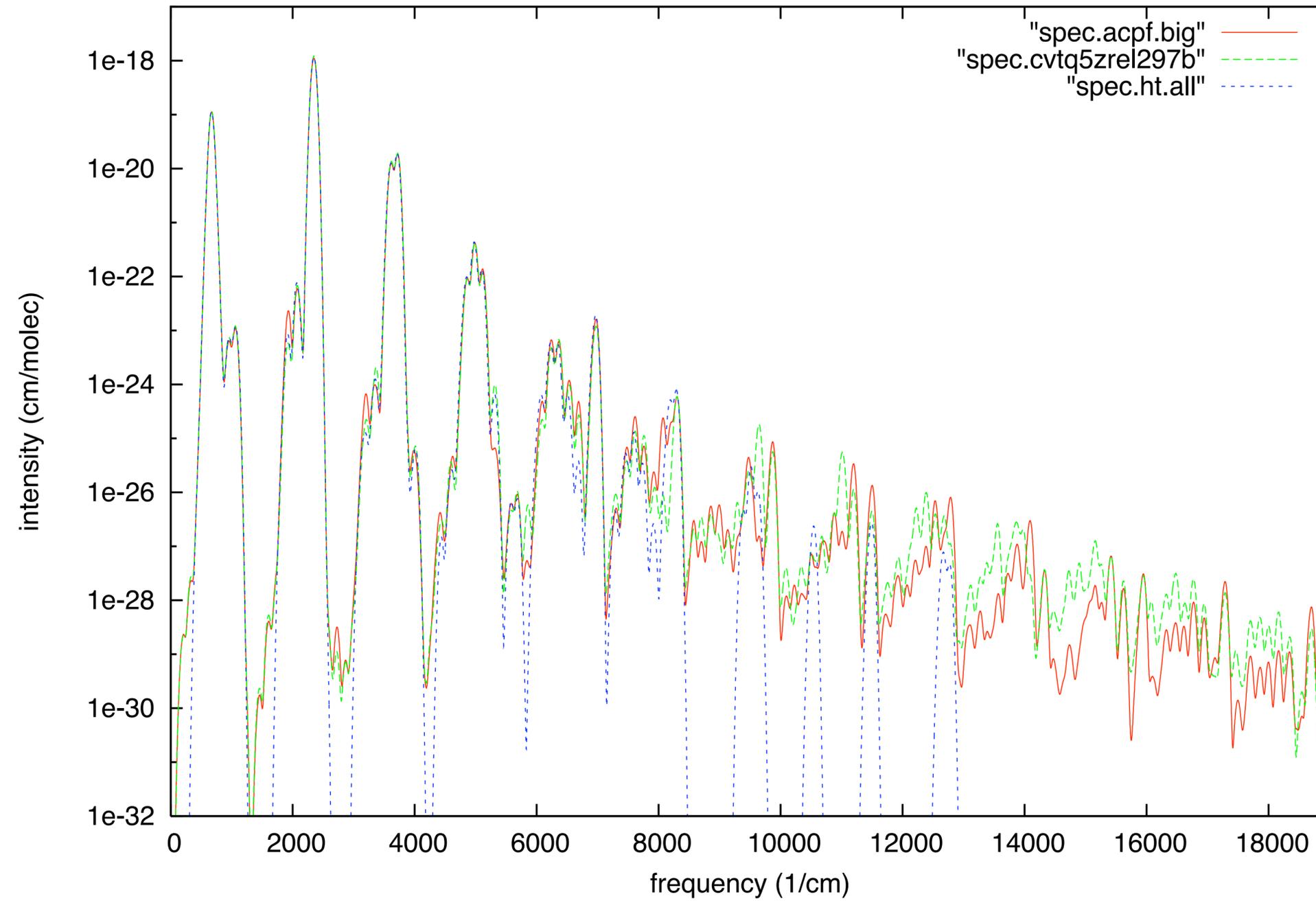


# Spectrum and Opacity of CO<sub>2</sub>. Comparison of Available Databases.

Richard Freedman<sup>1,2</sup> and David Schwenke<sup>2</sup>

(1) SETI Institute (2) Nasa Ames Research Center

- CO<sub>2</sub> is possibly of interest for irradiated, hot Jupiters and for extrasolar terrestrial planets
- Currently there are only a few databases available for high temperature CO<sub>2</sub>
- New calculations are under way by David Schwenke to produce an improved and expanded version of the CO<sub>2</sub> linelist

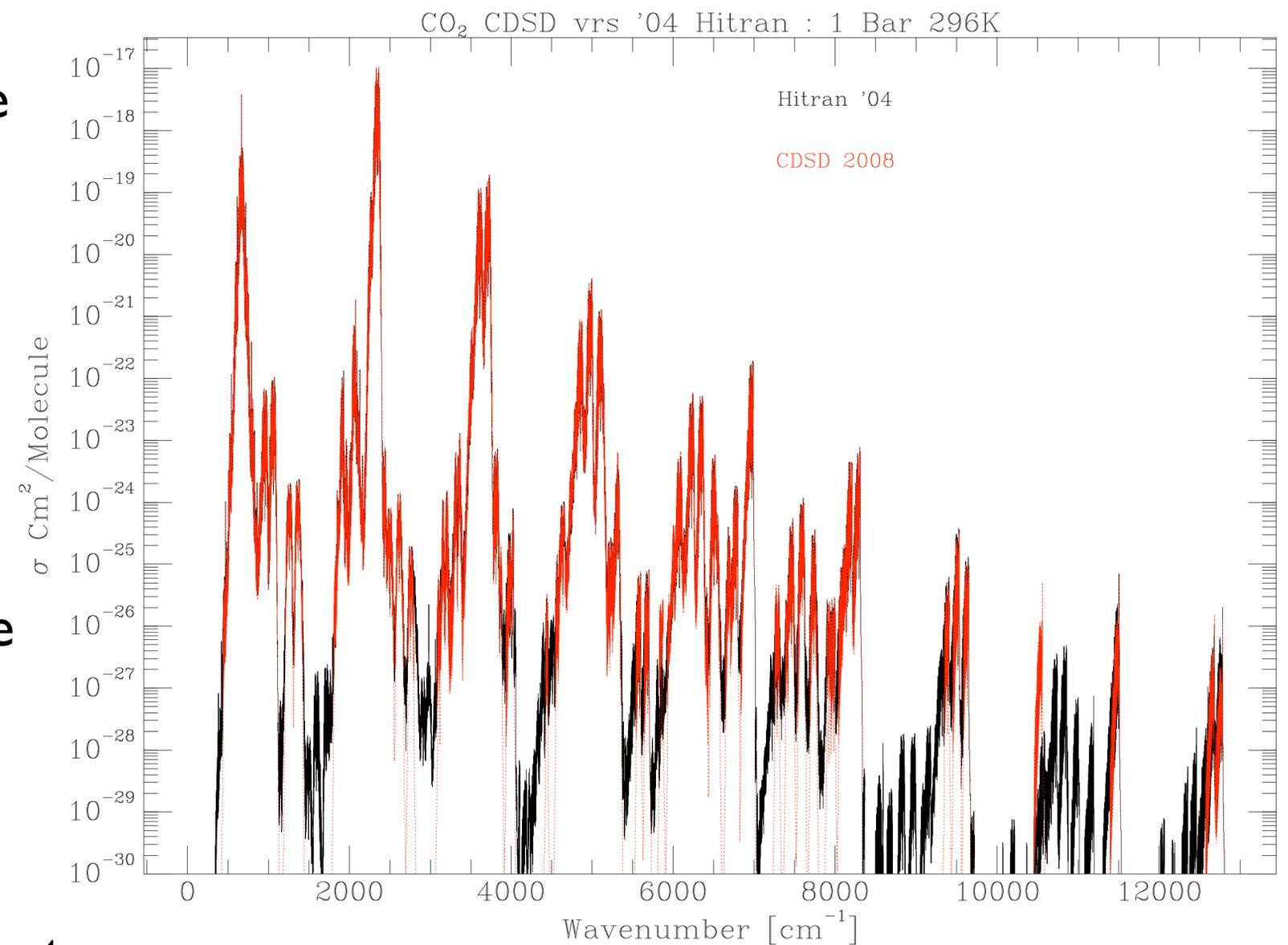


**A comparison of 3 different dipole moment functions for CO<sub>2</sub>**

- Hitran intensities are reproduced using the current Hitran database
- Two different dipole moment functions (DMF) are shown for CO<sub>2</sub> based on calculations by David Schwenke
- Note that the DMF from Schwenke extends to much higher wavenumbers than the current databases. This reflects the limits of the earlier potential energy surfaces
- A line list is expected by the end of 2008

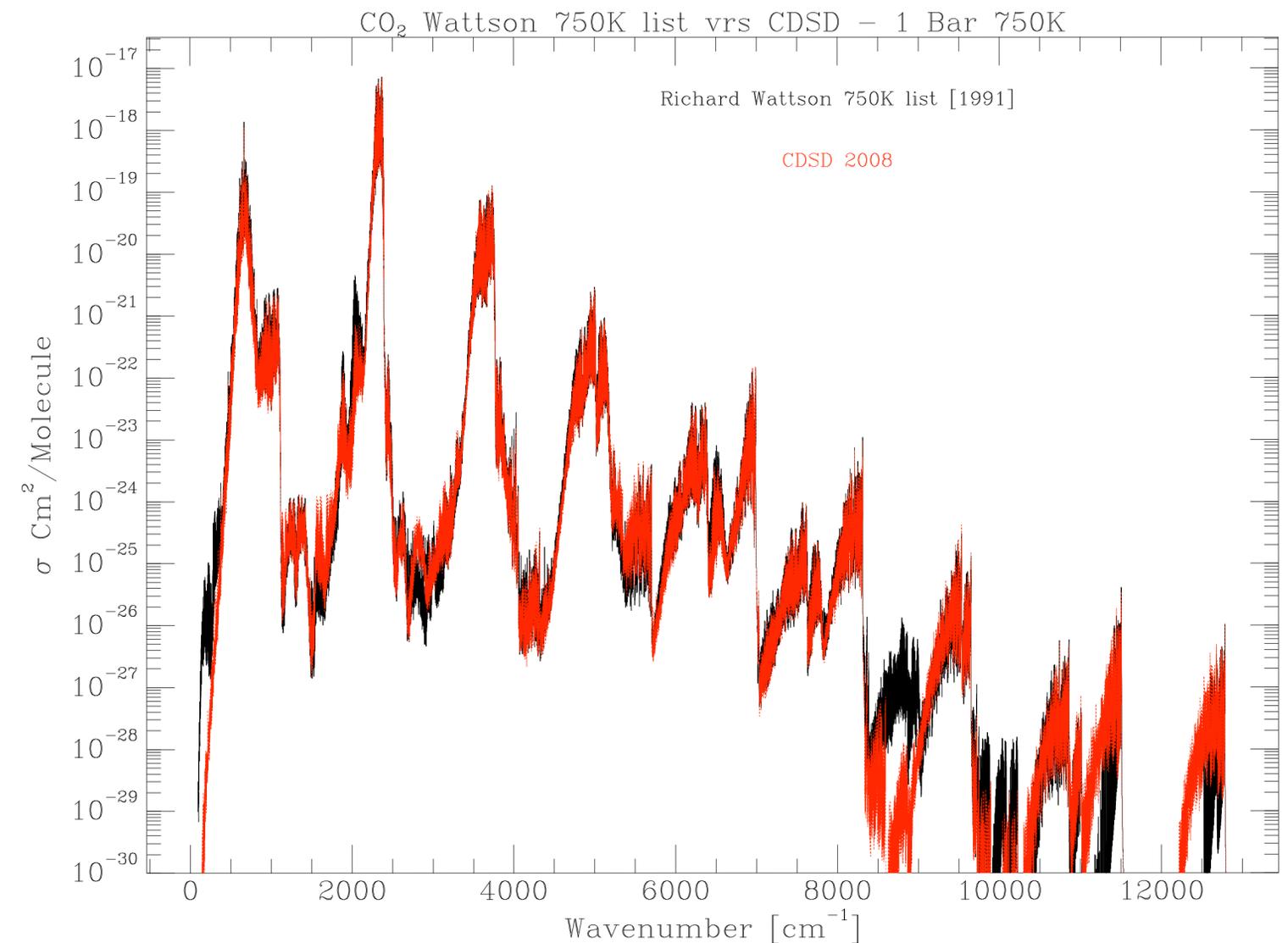
# Comparison of two CO<sub>2</sub> Spectra: Hitran 2004 vrs CDSD 2008

- Calculations are done using a Voigt profile with no adjustments ( $\chi$  factor) to the far line wings
- The line widths are appropriate for “foreign” (O<sub>2</sub> + N<sub>2</sub>) broadening
- The Russian (CDSD) data is missing some minor isotopes present in Hitran



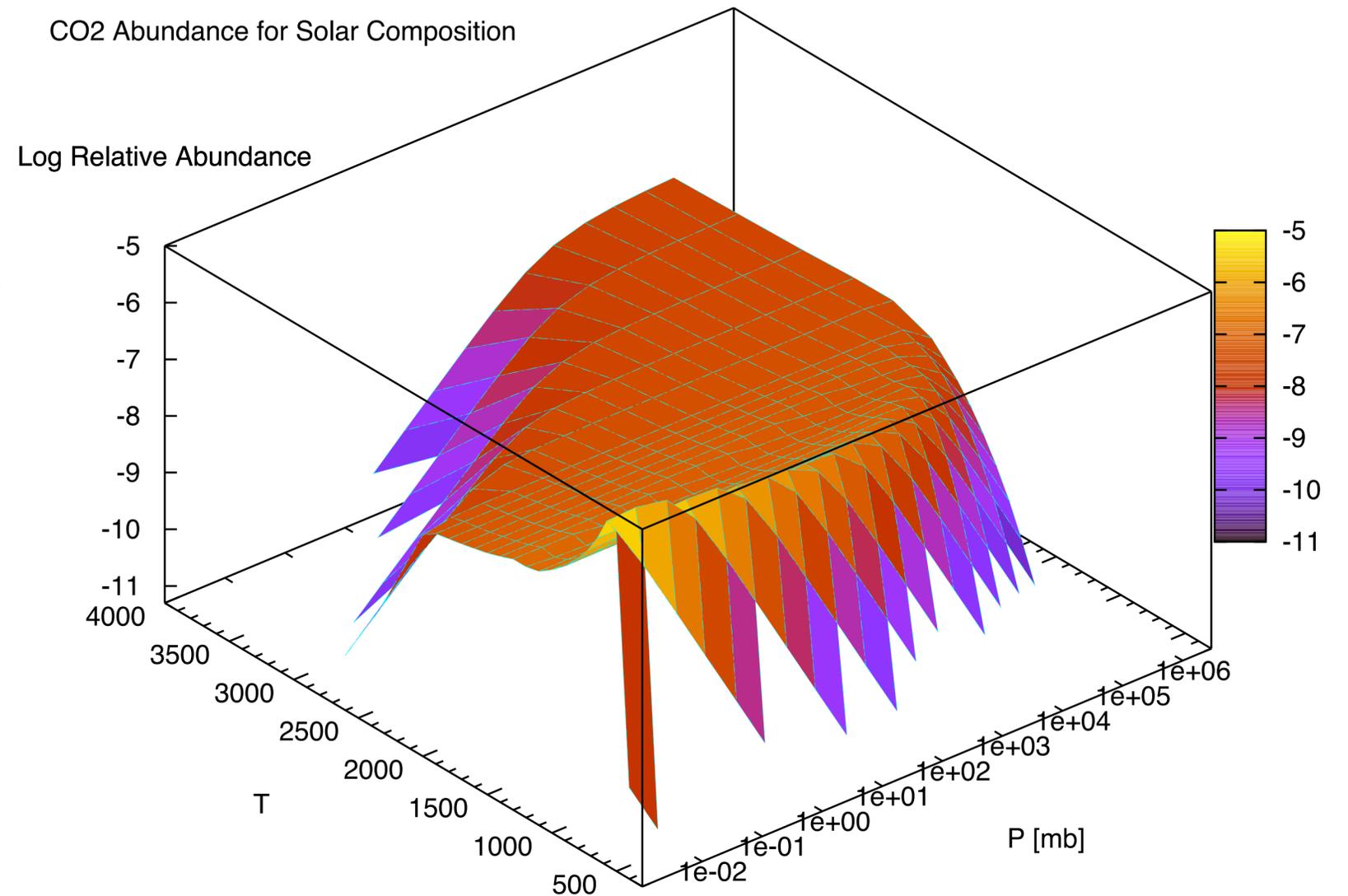
# CO<sub>2</sub> absorption at 750K & 1 Bar: Comparison of a 1991 calculation of Wattson and 2008 CDSD data

- Both calculations use a pure Voigt profile
- Assumes foreign broadening
- The Wattson data includes some bands added by Freedman at high wavenumbers
- The Hitran and Wattson data show better agreement as they share some common computational history (based on DND calculations and similar dipole moments)



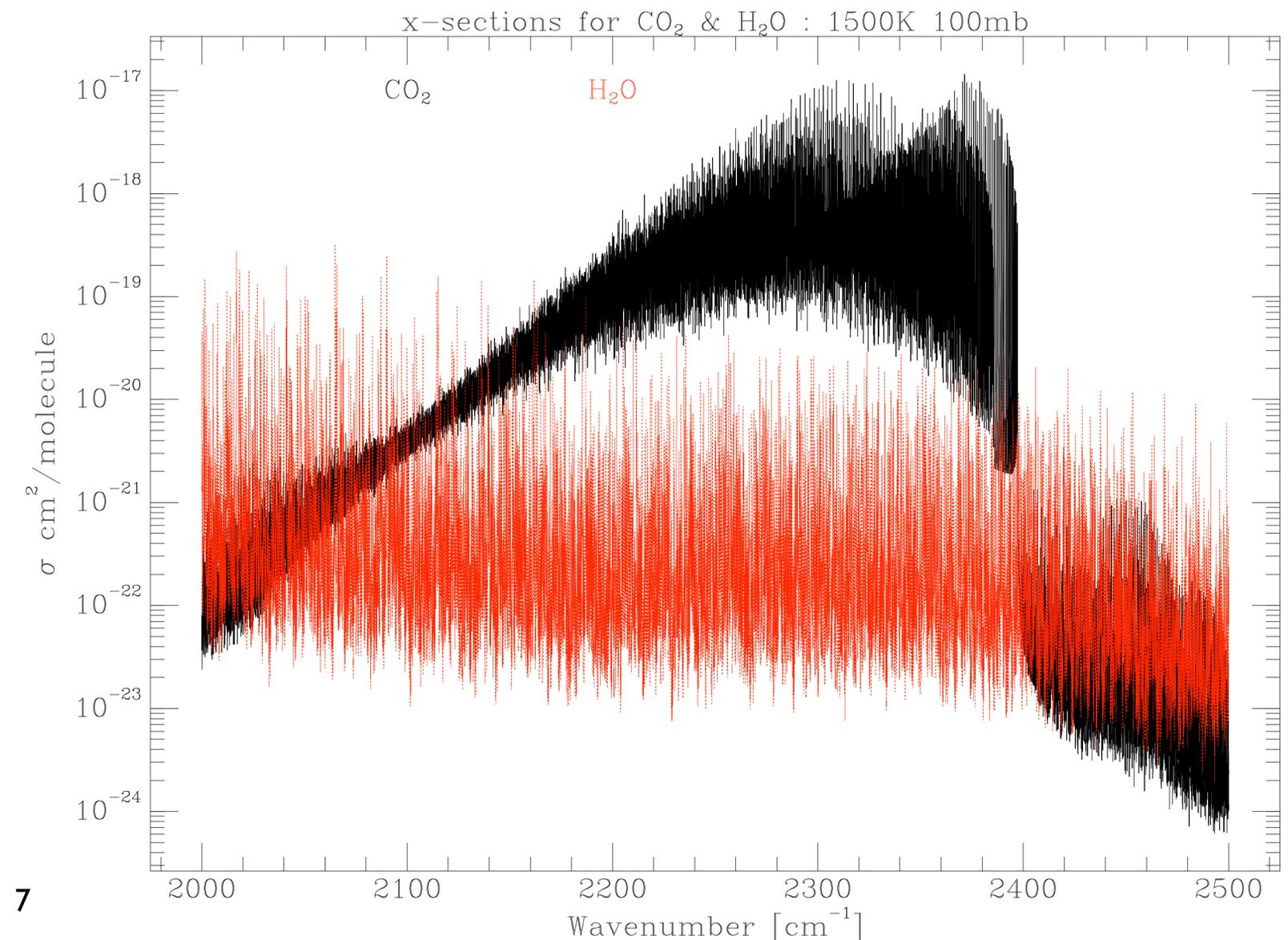
# Equilibrium abundances of CO<sub>2</sub> for an assumed solar composition

- In chemical equilibrium the most significant abundances are at moderate temperatures and low pressures
- For temperatures  $> \sim 1200\text{K}$  the largest abundances are at the higher pressures
- At high temperatures the abundance is greatest at high pressures
- Non-equilibrium abundances may be quite different in certain P:T regimes



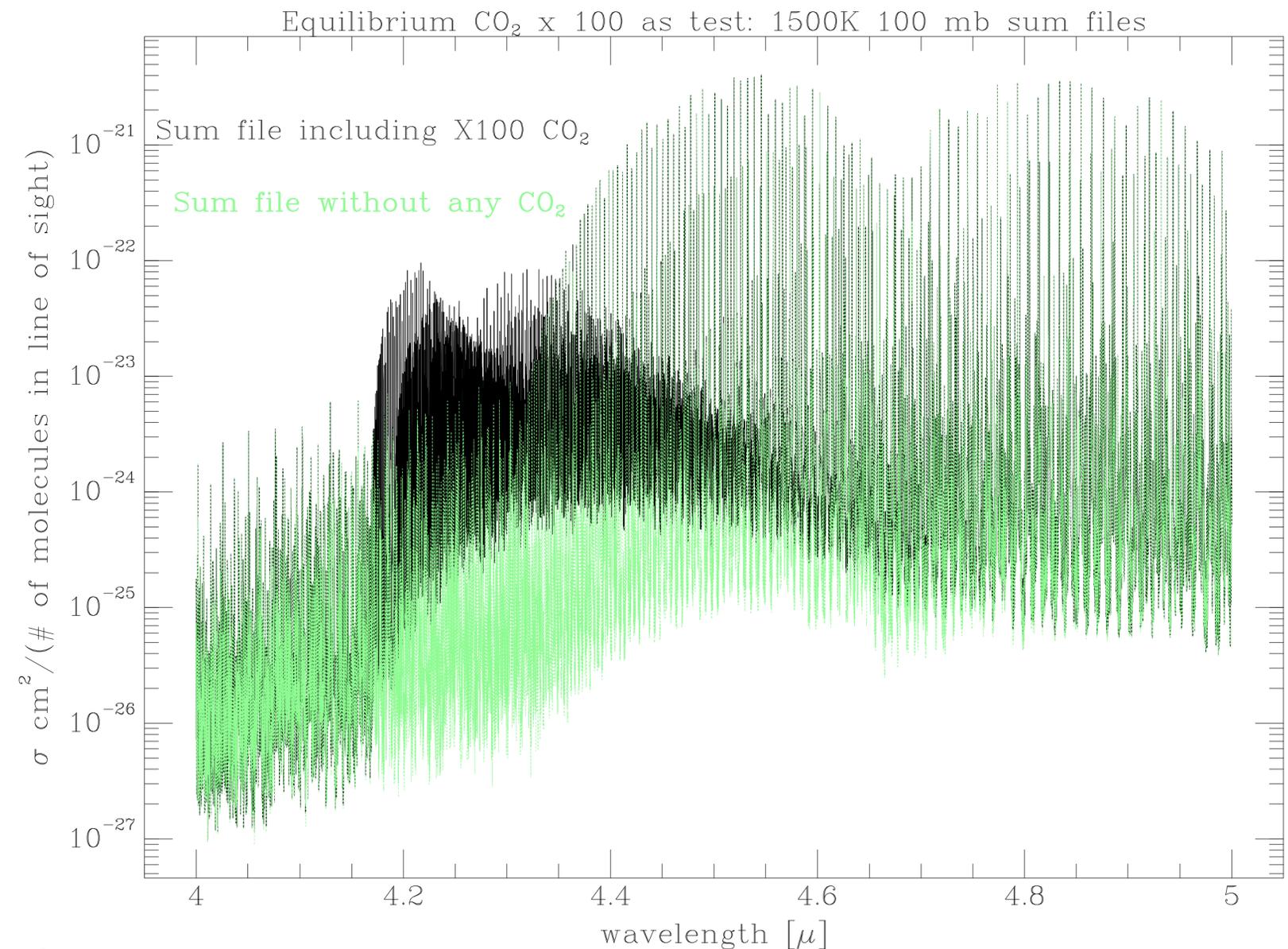
# Comparison of absorption cross sections for CO<sub>2</sub> & H<sub>2</sub>O

- Cross-sections are shown per molecule without any adjustment for the relative abundance
- The region shown is in the CO<sub>2</sub> 4.3 μm band



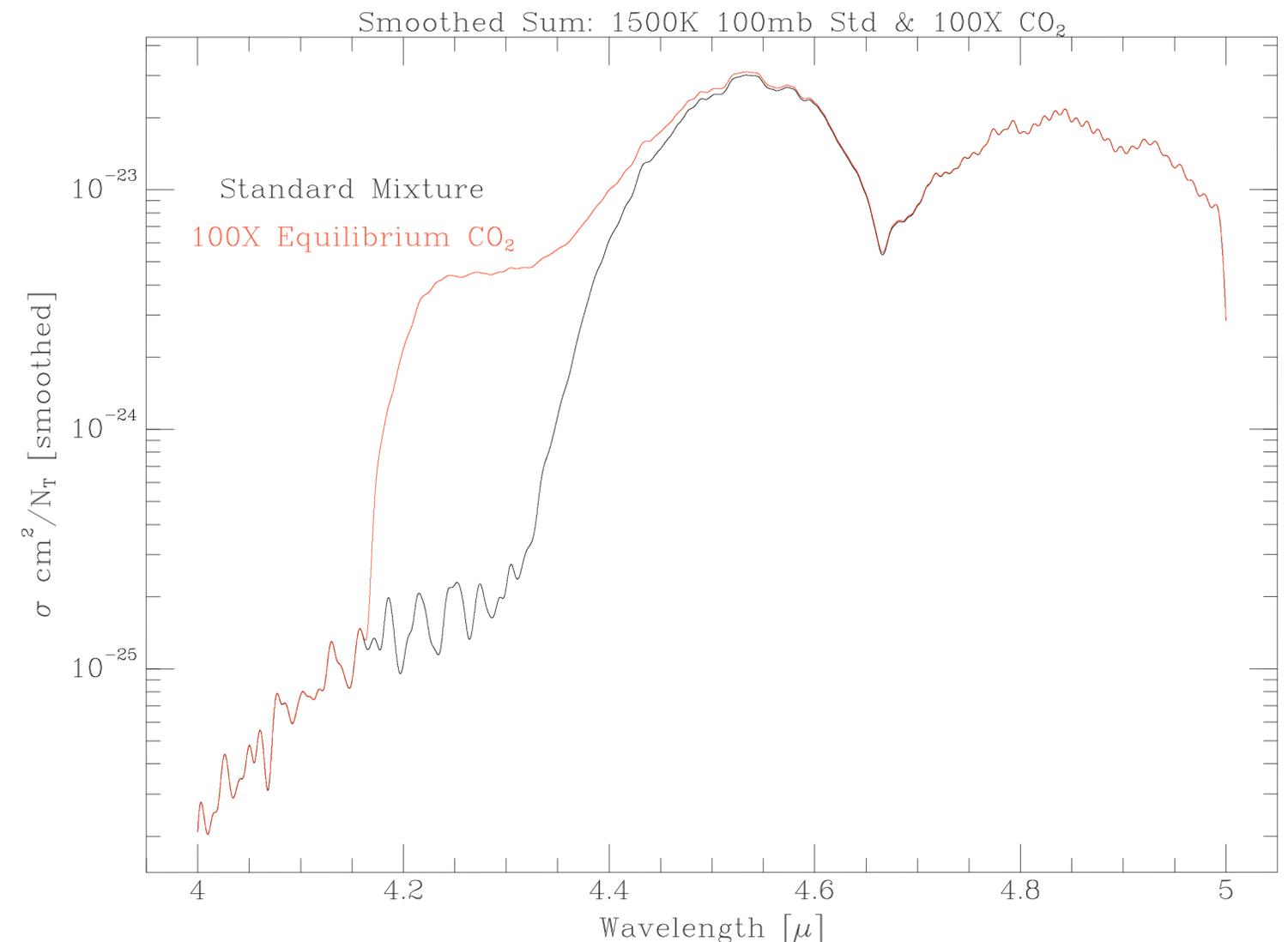
# Sum of opacities of 10 molecules and alkali atoms + CO<sub>2</sub> x 100

- For this test the relative abundance of CO<sub>2</sub> has been increased by a factor of 100 from the chemical equilibrium abundance
- This may be appropriate for upper regions of an exoplanet atmosphere affected by UV radiation from the central star
- This shows the raw absorption coefficients unsmoothed



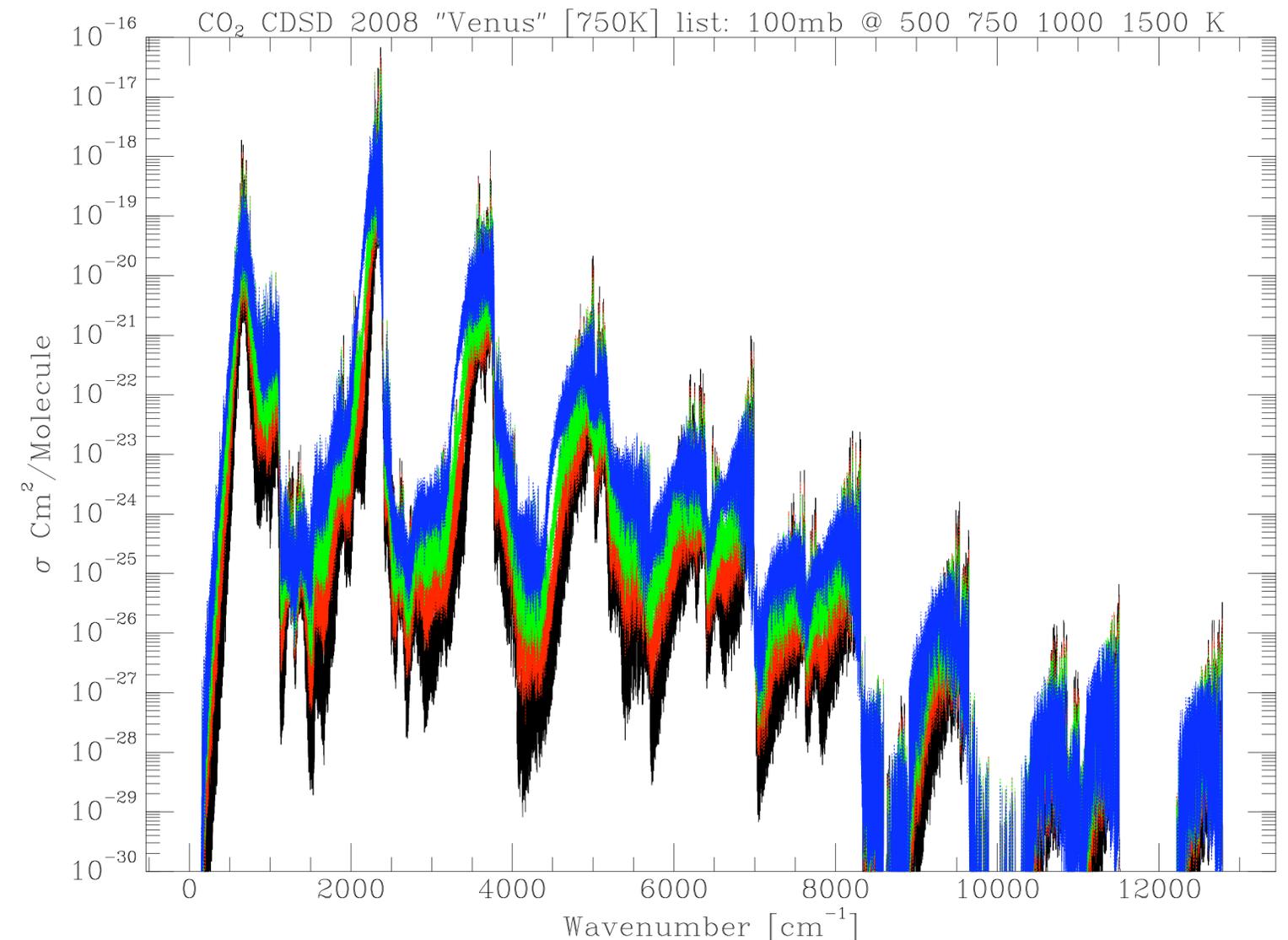
# Sum file smoothed to simulate an observation - resolution $\sim 1200$

- This plot shows more clearly the affect that enhanced  $\text{CO}_2$  abundance can have on the total opacity
- To calculate the total optical depth multiply the absorption cross-section by the total number of particles in the line of sight
- CIA has not been included [ $\text{H}_2\text{-H}_2$  and  $\text{H}_2\text{-He}$ ] as the total pressure is so low



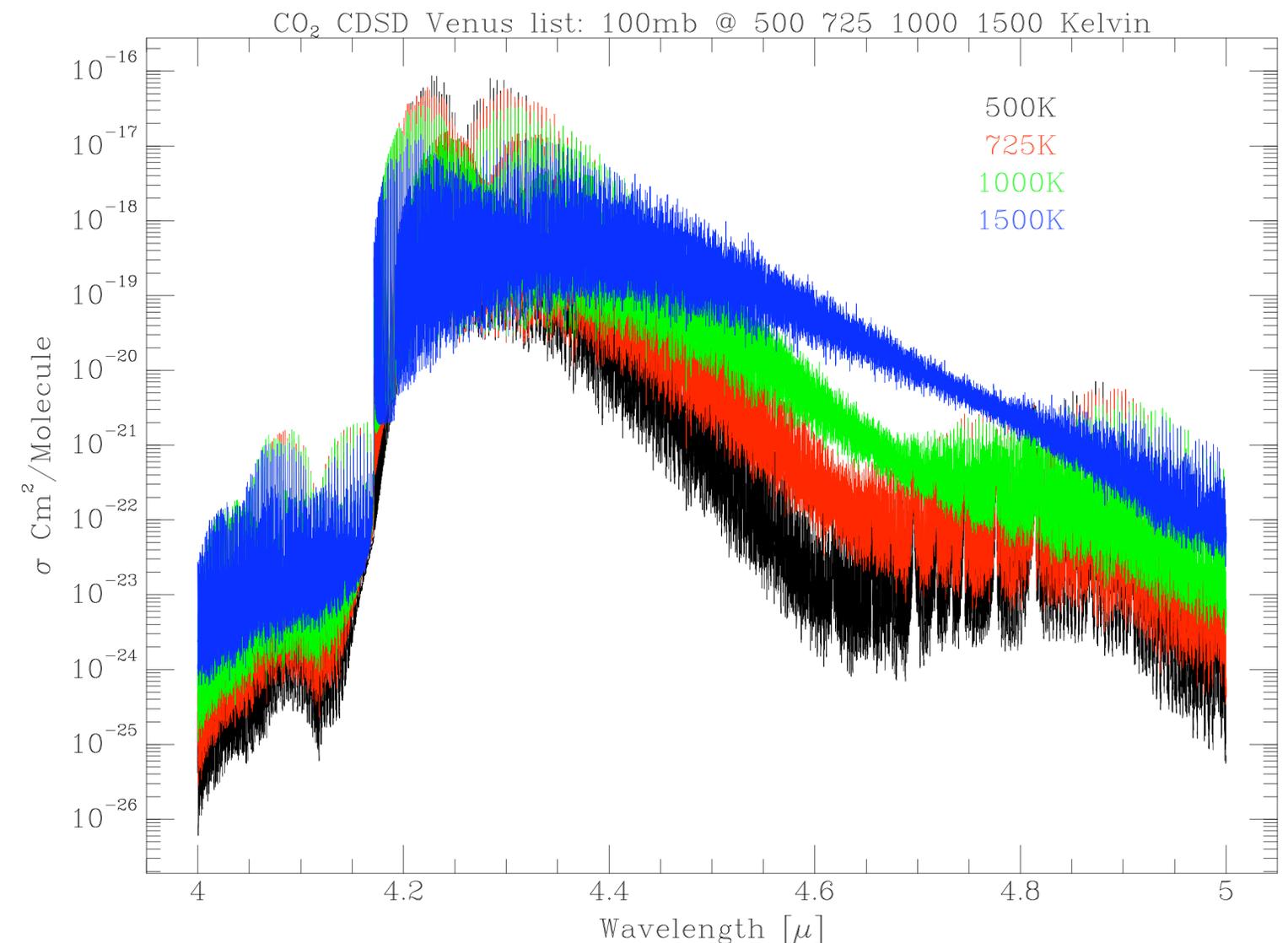
# Absorption coefficient of CO<sub>2</sub> for 4 different temperatures : 100 mb

- Classical Voigt profile was used with no adjustment for sub-Lorentzian wings of CO<sub>2</sub>
- Only the region around the 4.3  $\mu\text{m}$  band is being considered as that region is also near a minimum of H<sub>2</sub>O absorption and is therefore suitable for a search for the presence of CO<sub>2</sub>



# CO<sub>2</sub> in the 4.3 μ band at 4 temperatures

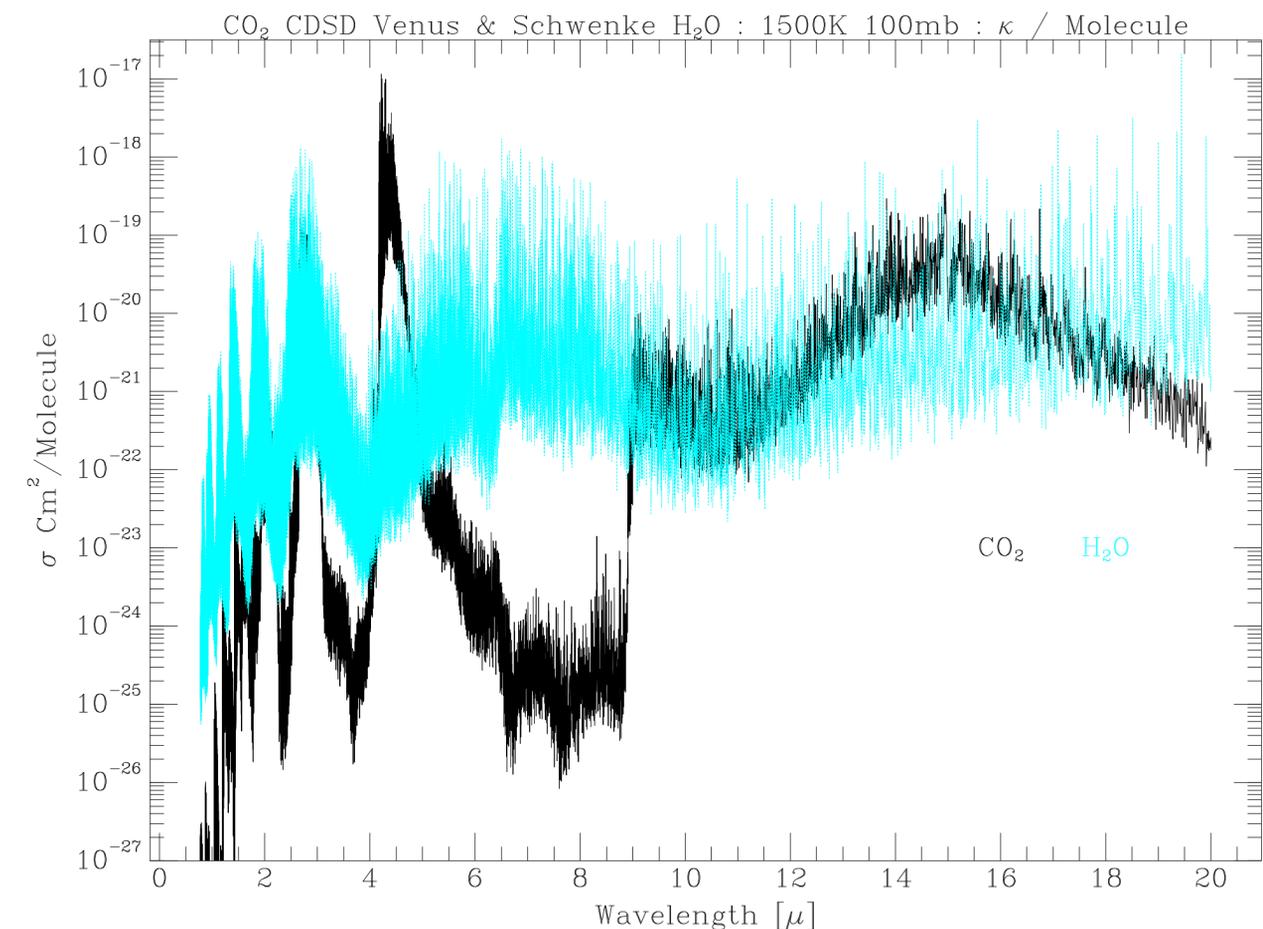
- Note that it is the weaker transitions which originate from excited energy levels that increase the most in a relative sense at high temperatures
- These calculations were applied to the file of the summed opacity of all molecules and the alkali atoms to show the effect of including CO<sub>2</sub> at a non-equilibrium abundance



- Because the 4.3  $\mu\text{m}$  band is opaque even at the operating altitude of SOFIA, observations from space will be required to observe in the 4.3  $\mu\text{m}$  wavelength region
- There is another region at  $\sim 1.05 \mu\text{m}$  which is near a minimum in the  $\text{H}_2\text{O}$  opacity that could be used to detect  $\text{CO}_2$  if its abundance were high enough. This region is also relatively transparent in the earth's atmosphere, however the intrinsic strength of  $\text{CO}_2$  absorption in this region is not really large enough for a detection unless the abundance greatly exceeds the equilibrium value and is much greater than the 100X case show here.

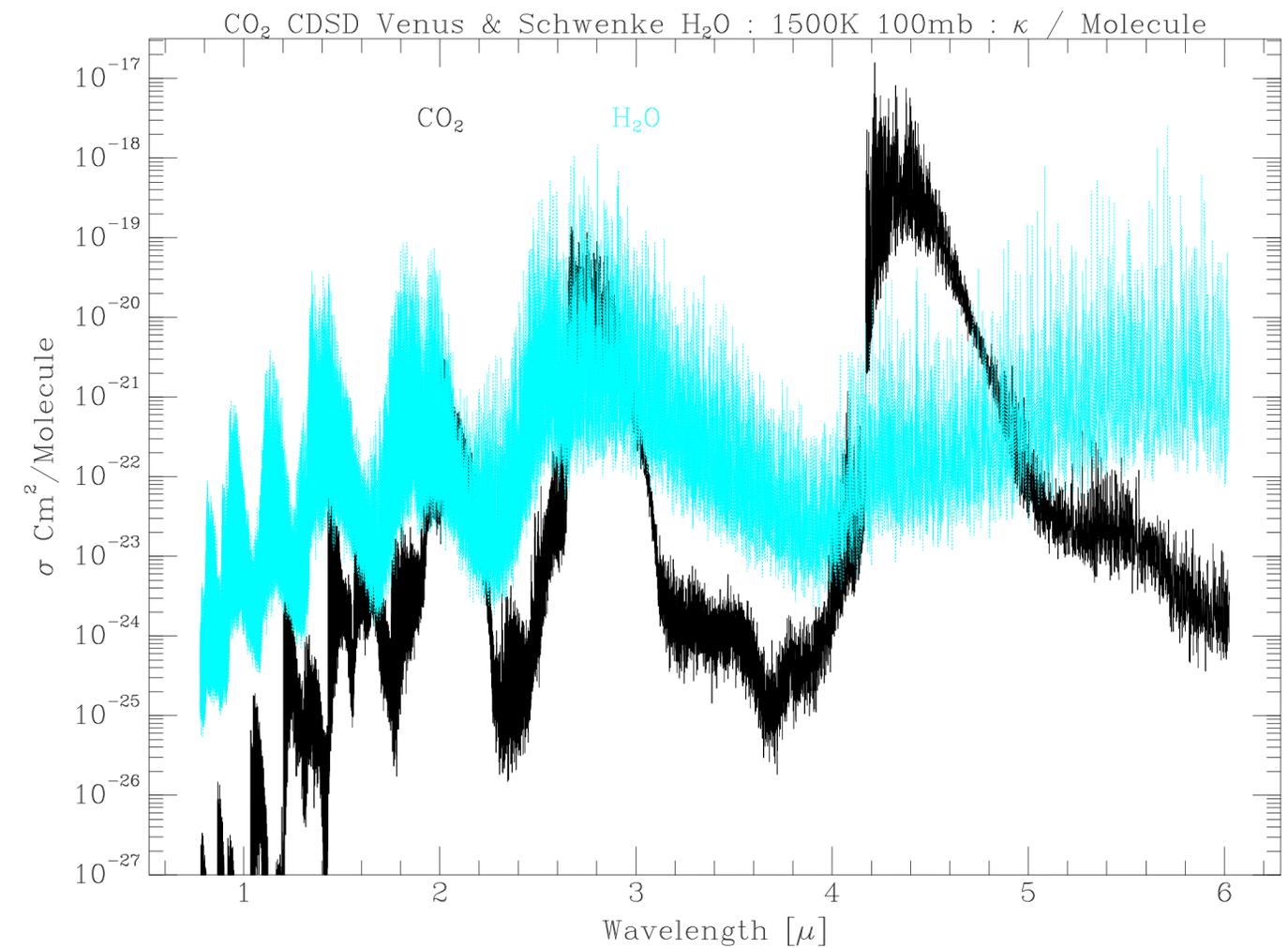
# Comparison of the opacity of CO<sub>2</sub> and H<sub>2</sub>O per molecule

- The 4.3 $\mu\text{m}$  band of CO<sub>2</sub> is the best placed to be detected in relation to the opacity of H<sub>2</sub>O which is a major source of opacity
- The 4.3 $\mu\text{m}$  cannot be observed from the ground because the atmosphere is opaque in this band even at aircraft altitudes



# A blowup of the short wavelength region

- Assuming equilibrium chemistry and solar abundances the relative abundance of CO<sub>2</sub> is less than that of H<sub>2</sub>O by at least a factor of 1000 in most of the regions where it exists in any appreciable quantity
- Non-equilibrium situations may allow detection of CO<sub>2</sub>
- Although the atmosphere is relatively transparent at  $\sim 1.05\mu\text{m}$  the CO<sub>2</sub> opacity there is still too low relative to H<sub>2</sub>O to be important in most cases



# Notes and References

- Russian CO<sub>2</sub> data: JQSRT 82 (2003) 165-196, CDSD-1000, the high temperature carbon dioxide spectroscopic databank, S.A. Tashkuna, V.I. Perevalova, J-L. Tebob, A.D. Bykova, N.N. Lavrentieva
- JQSRT 48 (1992) 537-566, Energy levels, intensities, and linewidths of atmospheric carbon dioxide bands, Rothman, L. S. and Hawkins, R.L. and Wattson, R.B. and Gamache, R.R.
- There are other articles about the techniques used by Wattson and Rothman in their calculations as applied to CO<sub>2</sub> & H<sub>2</sub>O

- The technique used by Wattson and Rothman is known as the DND or direct numerical diagonalization technique
- The methods used by D. Schwenke involve techniques closer to ab-initio quantum mechanical techniques