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Cosmic Dust Physical Properties and Laboratory Experiments



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Motivation Cosmic Dust Studies



- Dust extinction, polarization, spectroscopy, continuum emission as diagnostic tools
 (Optical depth, mass, magnetic fields, temperature, chemistry, growth processes and mixing, ...)
- Thermal, dynamical, and chemical structure
- Interesting structural and optical behaviour
 (Tunneling processes at low temperatures, vacuum cleaner, ...)

Photoelectric heating



Bakes and Tielens (1994)

The Facts

- Silicates and carbonaceous ISM dust
- Broad size distribution
- Additional materials in circumstellar envelopes

(carbides, nanodiamonds, fullerenes, ...)

- Molecular ices in cold clouds
- Grain growth in pp disks
- Crystalline silicates and molecular ices in pp disks



(For silicates: Henning, ARAA, 48, 21, 2010)

Dust emission spectrum



Fig. 4. Dust emission spectrum. Observations (crosses) pertain to the "cirrus" interstellar diffuse medium (see Table 1 and text). The horizontal bars represent the filter width used in the observations (given in Table 1). The model resulting spectrum (continuous line) is the sum of the three components that are PAHs, VSGs, and BGs.

Désert, Boulanger & Puget (1990) More to come: Compiègne et al. (2011)

Basic Types of Dust Mixtures



Dust in the Diffuse Galactic ISM



Whittet et al. 1997

See Chiar et al. 2000, Chiar & Tielens (2006), Van Breemen et al. (2011)

- No evidence for crystalline silicates in the diffuse ISM (<2%, e.g., Li & Draine 2001, Jäger et al. 2003, Kemper et al. 2004)
- Amorphization by cosmic rays/shock processing in ISM/ recondensation of amorphous silicates in the ISM (e.g. Jäger et al. 2003)
- 3.4 micron absorption feature Aliphatic hydrocarbons (e.g. Pendleton & Allamandola 2002)

Crystalline Revolution (ISO and Spitzer)



Jäger et al. (1998)

RECX5: Hale Bopp Formation around an M4 star?



IRS (5-40 μ m long slit, R=150, 10-38 μ m echelle, R=600)

Crovisier et al. (1997), see also Wooden et al. (1999, 2000)



Bouwman et al. (2010)

Comparison of the 10 µm Si-O stretch band



Spectral ambiguity

- A **GEMS in IDP L2011*B6**
- **B** Elias 16
- C Trapezium
- **D DI Cep (T Tauri star)**
- **Ε** μ Cep (M supergiant)

GEMS:

(Mg+Fe)/Si~0.7 (Keller & Messenger 2004) Mg/Si=0.6 and Fe/Si=0.4 (Ishii et al. 2008)

Bradley et al. (1999), Chiar & Tielens (2006), van Breemen et al. (2011)

IR Properties of Silicates – Amorphous vs. Crystalline Structures



IR Properties of Silicates – Amorphous vs. Crystalline Structures

- 10µm band due to Si-O stretching; position depends on level of SiO₄ polymerization (e.g. band shifts from 9.0 µm for SiO₂ to 10.5 µm for Mg_{2.4}SiO_{4.4} – Jäger et al. 2003)
- 18 µm band additionally broadened (coupling of the Si-O bending to the Me-O stretching vibration)
- Crystalline silicates: Bands beyond 20 µm caused by translational motion of metal cations within the oxygen cage and complex translations involving Me and Si atoms

Laboratory Investigations of Cosmic Dust





EELS – Fe (red), Mg (green), C (blue); J. Bradley/H. Ishii

MPIA Jena He droplet experiment

- Interplanetary dust particles and stardust in meteorites
- Optical properties of cosmic dust analogues
- Formation and modification of dust grains

Stardust in primitive meteorites and IDPs

Graphite	10 ppm	1-20 µm	Novae, SN, AGB
Diamond	1400	0.002	SN(?)
SiC	14	0.3-20	AGB (mainstream), SN
Al_2O_3	0.01	0.5-3	Red giants, AGB, SN
Si ₃ N ₄	0.002	1	SN

Onion-like presolar graphite particle -Murchison meteorite



Clayton et al.

Detection of nanodimanonds in unprocessed Allende



Banhart et al. (1998)

Silicates from Space





Scale bar: 200 nm

- 3 Olivine grains
 4 Pyroxene grains
- 3 Glass-like grains

Hoppe et al. 2005

(see also Messenger et al. 2003)

SEM Image

Why does interstellar dust exist?



- Dust destruction in diffuse ISM more efficient than production by AGB stars (see Jones & Nuth 2011)
- SN dust production rate seems to be very low
- "Homogeneous" dust models (Draine & Lee) vs. core-mantle models (Greenberg) vs. "inhomogenous dust" (Mathis)
- What is the nature of the VSG?
- Why don't we see SiC grains in the diffuse ISM?

Grain Sizes – From "Nano to Micro"

Coronene

Carbon Onion

Soot Particle







36 atoms 1 nm 10⁵ atoms 15 nm

10⁷ atoms 200 nm

Formation of Dust

Grain formation experiments under high-T conditions



Jäger ea. 09

HT (≥3500 K): Very small fullerene-like carbon grains LT (≤1700 K): Synthesis of PAH-based structures

Grain formation experiments under low-T conditions

Nuth & Moore (1989): Silicate material from molecular precursorsDartois et al. (2005):Formation of HAC polymers produced by UV
photolysis at low T

Transition from Carbon Clusters to Solid Particles



Non-crystalline disordered carbons

Soot Particles (without hydrogen/oxygen): Curved and closed structures or polycrystalline materials Soot Particles (with hydrogen): Smaller grains preferably formed: Curved structures





Arc discharges, laser ablation, thermal sublimation methods, sputtering, laser pyrolysis, combustion

Gas-phase condensed soot particles



Grain formation at high temperatures



Jäger ea. 09

HT (≥3500 K): Very small fullerene-like carbon grains LT (≤1700 K): Synthesis of PAH-based structures



Characterization of the PAHs in LT condensate

<u>Matrix-Assisted-Laser Desorption</u> and <u>Ionization combined with</u> mass spectrometry MALDI-TOF

PAHs with masses 12000 up to 3000 Da were found 8000 Intensity 4000 $C_{222}H_{42}$ 0 500 1500 2000 2500 3000 0 1000 Masses in Da

Soot formation pathways HT Condensation Process T ≥ 3500 K

Fullerene-like carbon seeds & fullerenes







Haberland, Clusters of Atoms and Molecules I, Springer Verlag



Dust and Radiation



Incoming radiation

- plane waves
- polarised somehow
- some spectrum

Absorption



- Transformation of energy to some other form
- Re-emission at different wavelengths

Scattering (elastic)

- Change in direction
- Change in polarization
- No change in wavelength

Let us construct a model ...



- 1. Assume chemical composition, shape, size, internal structure distribution
- 2. Select the relevant laboratory data for n, k (material structure? temperature?)
- 3. Calculate the cross sections (scattering codes)
- 4. Construct appropriate mean values
- 5. Apply these data in your radiative transfer calculation (or simple fitting procedure)

Interstellar Dust Grains – Opportunity and Challenge



"It is a difficult experimental task to produce particles a few hundred angstroms in size, keep them completely isolated from one another and all other solids, maintain them in ultra-high vacuum and at low temperatures, and study photon interactions with the particles from far infrared to extreme ultra-violet. This is the opportunity we have in the case of interstellar dust."

Donald D. Huffman Adances in Physics, 26, 129 (1977)

Basic Optical Properties of Solid Particles



Basic Optical Data Cosmic Dust Analogues



- Broad Wavelength Range
- Appropriate Structure (Fe/Mg, am./cryst. ...)
- Isolated Small Particles
- Temperature Range



MPIA Lab Astrophysics Group at the University of Jena

Heidelberg-Jena-Petersburg database of optical constants (Henning et al. 1999)

http://www.mpia-hd.mpg.de/HJPDOC/

Optical behaviour of small particles



After Krügel (2003, p.235) – Absorption (dots), extinction (solid line)

What you need to know ...



What you need to know ...





- Interstellar UV bump
- Near-infrared extinction properties
- Far-infrared absorption properties

Origin of the Strong UV Resonance



- Remarkable constancy of peak position (4.60 μm⁻¹; variations smaller 1%)
- Peak width varies around mean value of 1.0 μm⁻¹ (variations smaller 25%)
- Lack of correlation between variation of peak position and width (except for the widest bumps: systematic shift to larger peak wavenumbers)
- Strength of the feature requires abundant element as part of the carrier
- Feature is pure absorption feature

What is the carrier?

• HAC nanoparticles (e.g. Schnaiter et al. 1998, Gaballah et al. 2011)

• Large PAHs (e.g. Beegle et al. 1997, Steglich et al. 2010)



Near-infrared Extinction Law



Optical Data of Amorphous Silicates: Mg_xFe_{1-x}SiO₃

Increase of NIR absorptivity with Fe content



(J. Dorschner, B. Begemann, Th. Henning, C. Jäger and H. Mutschke, A&A 1995)

What are the FIR Properties of the materials?

- Structural composition of the material (e.g. Jäger et al. 1998)
- Grain size and agglomeration state (e.g. Henning & Stognienko 1996)
- Temperature of the material to be discussed



FIR Absorption Efficiency/Spherical Particles



FIR Absorption Efficiency/CDE



Extinction Spectra of Carbonaceous Materials



Quinten, Kreibig, Henning, Mutschke (2002)

LOW-TEMPERATURE EFFECTS

Henning and Mutschke (1997)

Crystalline Dielectric Solids

- IR bands (single phonon transitions): Sharpening because of decreased damping, shift to shorter wavelengths
- FIR absorption (phonon difference processes): significant reduction because of decreasing phonon number

Amorphous Dielectric Solids

- FIR absorption: Dominated by disorder-induced single phonon processes,
- no temperature dependence
- Millimeter range: highly temperature-dependent low energy processes, e.g. tunneling transitions in glasses

Semiconductors

• free charge carrier absorption: vanishes because conduction band is depopulated

What is expected ?

=>Bands are broadened and shifted to lower frequencies with higher temperature



How big is the relative peak shift ?

Long-wavelength forsterite bands as thermometer

Koike et al. (2006)

Temperature Dependence – Laboratory Studies

- **Bösch (1978):** silica glass (500 μm – 5 mm, 1.2 – 300 K)
- Agladze et al. (1996): crystall. and amorphous silicates $(700 \ \mu m 3 \ mm, 1.2 30 \ K)$
- Mennella et al. (1998): am. carbon, crystalline silicates, am. fayalite (20 µm – 2 mm, 24 – 294 K)
- Boudet al. (2005): am. silica, am. silicate ($100 \ \mu m - 2 \ mm, \ 10 - 300 \ K$)

Between 500 µm and 2 mm: Anticorrel. between T and B

Am. silicate grains with olivine composition (Mg_xSiO₄)

Experiments by K. Demyk et al.

Which new dust features can we expect to see with Herschel?

FIR: Lattice vibrations of heavy ions or ion groups with low bond energies (example KBr: transverse optical mode at 86 μ m); PACS: 57-210 μ m

- Forsterite 69 µm band
- Fayalite 93-94 µm and 110 µm band
- Crystalline Diopside $65-66 \ \mu m$
- Hydrous silicates 100-110 μm (e.g. montmorillonite)
- Calcite CaCO₃ 92 µm

Herschel – Predictions and PACS Spectra

HD 100546, DIGIT Program
Sturm, Bouwman, Henning et al. (2010)Measured position is 69.2 μmSturm, Bouwman, Henning et al. (2010)(Cold (50 K) iron-free forsterite has a peak at 69.0 μm)

- a) Warm iron-free grains create the shift (150-200 K) (Mulders et al. 2011)
- **b)** Cold forsterite with a few percent iron shifts feature

FIR optical data of interesting materials

Carbonates, silicates, hydrous silicates at low temperatures

- Hydrous silicates: Mutschke
- Carbonates:
- Olivine:
- Diopside:

Mutschke et al. (2008) Posch et al. (2007) Bowey et al. (2001), Suto et al. (2006), Koike et al. (2006) Bowey et al. (2001), Chihara et al. (2001)

 $CaMg[Si_2O_6]$

Towards a Dusty Universe

ALMA

JWST

- Basic understanding of grain properties
- Diversity of grain properties in galaxies
- Formation and evolution of grains Next challenge